Year 5 Autumn 1 Science- States of Matter

Key facts

- The particles in a solid are rotating, vibrating or moving about a fixed point, close to each other.
- The molecules in a liquid move more and have more energy than particles in a solid but remain in close contact with each other.
- The particles in a gas have a lot of energy, moving around in a random way, hitting other particles and the walls of the container.
- Evaporation is when heat is applied to a liquid the particles move around even faster, breaking away from each other and forming a gas or vapour.
- 5. **Condensation is when** a gas has heat removed from it; the particles in the vapour slow down, come together and loosely bond, returning to the liquid state.

